Tetrahedron Letters 50 (2009) 6944-6946

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



A facile and regioselective synthesis of 2,5-disubstituted pyrroles via gold-catalyzed cycloisomerization of acetylenylaziridines

Dong-Dong Chen, Xue-Long Hou*, Li-Xin Dai

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, China

ARTICLE INFO	A B S T R A C T
Article history: Received 13 May 2009 Revised 14 May 2009 Accepted 15 May 2009 Available online 28 May 2009	Gold-catalyzed cycloisomerization reaction of acetylenylaziridines provides 2,5-disubstituted pyrroles in high yields. The presence of protic species accelerates the reaction rate and improves the yields of pyrrole products. © 2009 Elsevier Ltd. All rights reserved.

Transition-metal-catalyzed cyclizations and cycloisomerization reactions of envnes have so far been a convenient protocol to render a variety of functionalized ring structures.¹ Among them the gold catalysts have demonstrated their unique catalytic activities through the activation of alkynes, emerging as the powerful catalyst for this rapidly developing research area in past few years.² The transformations of epoxides³ and cyclopropanes⁴ containing carbon-carbon triple bond to furans, polycyclic ring systems and other functionalized compounds in the presence of gold compounds as catalyst have been reported. However, few reports were given on the gold-catalyzed transformation of aziridines in spite of the fact that the metal-catalyzed conversions of aziridines have well been documented.⁵ We have developed several simple procedures for the synthesis of functionalized aziridines including acetylenylaziridines.⁶ We envisage that such triple bond in acetylenylaziridines could be activated by gold compounds and the pyrrole formation⁷ seems to be feasible via intramolecular attack of nitrogen of the aziridine towards gold(I)-activated triple bond. Herein, we present a new synthetic approach to 2,5-disubstituted pyrroles utilizing the acetylenylaziridines in the presence of Au catalyst.⁸

The choice of catalyst to activate the nitrogen of aziridine with low nucleophilicity is vital,^{7a,9} although it was easy with which the alkynyloxiranes were isomerized to furan with gold(III) chloride as a catalyst via the attack of oxygen of epoxide as a nucleophile to the activated triple bond.¹⁰ Thus, different metal salts and complexes were screened by using aziridine **1a** for our initial investigation (Eq. 1, Table 1). In the presence of 10 mol % of transition-metal complexes IPrCuCl and Ru(PPh₃)₂Cl₂ or metal salts Y(OTf)₃, In(-OTf)₃, PtCl₂¹¹ and AgOTf, the formation of pyrrole **2a** was not observed (entries 1–6). However, a few percentage of pyrrole **2a** was detected when Au compounds were used (entries 7, 8 and 10). The yield of **2a** increased dramatically when 10 mol % of AgOTf

* Corresponding author. E-mail address: xlhou@mail.sioc.ac.cn (X.-L. Hou). was added (entries 11, 12, 14 and 15).¹² The presence of protic species plays an important role. Good yield of **2a** was provided if 0.1 mL of MeOH was added to the reaction system (entries 14 and 15) while lower yield of **2a** was furnished without the use of MeOH (entry 11).¹³ It is noticeable that removing trace amount of water from the system with 4 Å MS prohibited this reaction (entry 13). The reaction still proceeded smoothly if the catalyst loading decreased to 5 mol %, providing 60% yield of **2a** in 17 h (entry 15).

$$\begin{array}{c} \text{Ph} & \begin{array}{c} \text{Is} & \\ \text{THF, r.t.} & \text{Ph} & \\ \text{THF, r.t.} & \text{Ph} & \\ \text{Ts} & \\ 1a & 2a \end{array}$$
(1)

Then we investigated the influence of solvents on these isomerizations (Eq. 2, Table 2). We found that the pyrrole **2a** was afforded efficiently if the reaction ran in toluene, DCM and THF (entries 1– 3), among them THF was a better choice, while poor results were provided if CH_3CN and CH_3OH were used as solvents (entries 4 and 5).

$$\begin{array}{cccc} Ts & & PPh_{3}AuCl (10 \text{ mol }\%) \\ & & \underline{AgOTf (10 \text{ mol }\%)} \\ & & & \underline{AgOTf (10 \text{ mol }\%)} \\ & & & \underline{Solvent/MeOH, r.t.} \end{array} \begin{array}{c} Ph & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & & \underline{N} \\ & & & \underline{N} \end{array} \begin{array}{c} Ph \\ & & \underline{N} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \\ & & \underline{N} \end{array} \begin{array}{c} Ph \\ & \underline{N} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \\ & \underline{N} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \\ & \underline{N} \end{array} \end{array} \begin{array}{c} Ph \end{array} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \end{array} \begin{array}{c} Ph \end{array} \begin{array}{c} Ph \end{array} \end{array}$$
 \end{array}

With the optimized conditions in hand, we have probed the scope of this isomerization reaction (Eq. 3, Table 3). Varying the aryl and alkyl substituents at either aziridines ring or the terminal carbon of triple bond had a little influence on the yield of the reaction, all reactions proceeded smoothly at room temperature and were completed within 10 h to afford 2,5-disubstituted pyrroles in high yields, except that of aziridine **1j** with Phenyl and acetyle-nyl group in 19 h (entry 10). High yields (87–99%) of pyrroles **2** were furnished when one substituent or two substituents on the aziridines ring and the terminal carbon of triple bond were alkyl (entries 4–8 and 13–14) while the yields were a little bit lower



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 Table 1

 Screening of metal catalysts in the cycloisomerization of aziridine $1a^a$

Entry	Catalyst	Time (h)	Yield ^b (%)	
1	IPrCuCl	48	N.R.	
2	Y(OTf) ₃	48	N.R.	
3	In(OTf) ₃	48	N.R.	
4	PtCl ₂	16	N.R.	
5	$Ru(PPh_3)_2Cl_2$	24	N.R.	
6	AgOTf	16	N.R.	
7	NaAuCl ₂ ·H ₂ O	96	8	
8	AuCl ₃	72	<5	
9	PicAuCl ₂ ^c	48	N.R.	
10	PPh ₃ AuCl	48	<5	
11	PPh ₃ AuCl/AgOTf	36	58	
12	AuCl ₃ /AgOTf	4	50	
13	PPh ₃ AuCl/AgOTf ^d	20	N.R.	
14	PPh ₃ AuCl/AgOTf ^e	9	69	
15	PPh ₃ AuCl/AgOTf ^f	17	60	

^a Reactions were conducted with a solution of **1a** (74.7 mg, 0.2 mmol) and 10 mol % catalyst (0.5 mL 0.04 M AgOTf–THF solution was used if necessary) in THF (0.2 M) protected from light at rt.

^b Isolated yield.

^c Pic = pyridine-2-carboxylato.

^d 4 Å MS was added.

^e MeOH (0.1 mL) was added.

 $^{\rm f}\,$ 5 mol % catalyst was used with MeOH (0.1 mL).

Table 2

Influence of different solvents on the cycloisomerization of aziridine 1a^a

Entry	Solvent	Time (h)	Yield ^b (%)
1	THF	9	69
2	Toluene	4	63
3	DCM	4	67
4	CH ₃ CN	6	N.R.
5	CH₃OH	6	Complex

^a Reactions were carried out as follows: **1a** (74.7 mg, 0.2 mmol) and PPh₃AuCl (10 mg, 0.02 mmol) with AgOTf–THF solution (0.5 mL, 0.04 M, 0.02 mmol) in solvent (0.5 mL) and MeOH (0.1 mL) protected from light at rt.

^b Isolated yield.

 Table 3

 Scope of gold(I)-catalyzed cycloisomerization of aziridines 1^a

		Substrate 1			Time (h)	Yield ^b (%)
Entry		R ¹	R ²	R ³		
1	а	Ph	Ph	Ts	9	69 (2a)
2	b	o-ClC ₆ H ₅	Ph	Ts	6	86 (2b)
3	с	1-Nap	Ph	Ts	2	74 (2c)
4	d	i-Pr	Ph	Ts	5	95 (2d)
5	e	Су	Ph	Ts	6	94 (2e)
6	f	n-Pr	Ph	Ts	3	99 (2f)
7	g	Ph	Ph	Ms	10	76 (2g)
8	h	Су	TMS	Ts	3	95 (2h) ^c
9	i	Ph	TMS	Ts	12	70 (2i) ^c
10	j	Ph	Н	Ts	19	59 (2i)
11	k	Ph	c-Pr ^d	Ts	3	94 (2k)
12	1	1-Nap	n-Bu	Ts	2	71 (2I)
13	m	Су	n-Bu	Ts	4	94 (2m)
14	n	Су	CH ₂ OCH ₃	Ts	3	87 (2n)

^a Reactions were carried out as follows: **1** (0.2 mmol) and PPh₃AuCl (10 mg, 0.02 mmol) with AgOTf–THF solution (0.5 mL, 0.04 M, 0.02 mmol) in mixed solvents of THF/MeOH = 5/1 (0.6 mL) protected from light at rt.

^b Isolated yield.

^c $R_2 = H$ in the product.

^d c-Pr = cyclopropyl.

(59–74%) if both substituents are aryl (entries 1, 3 and 7) or aryl and TMS or H (entries 9 and 10) with the exception of the reaction of **1b** and **1l** (entries 2 and 12).



Scheme 1. Mechanistic hypothesis for the cycloisomerization of aziridines.

$$\mathbb{R}^{1} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{2} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{2} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{2} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{2} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{2} \xrightarrow[\mathbf{R}^{2}]{} \mathbb{R}^{3} \xrightarrow[\mathbf{R}^{2}]$$

The reaction seems to be like that reported in the literature by Hashmi and Sinha¹⁰ to proceed via the gold(I)-activation of alkyne followed by cycloisomerization to produce intermediate **A**, deprotonation of it affords intermediate **B**, which is converted to pyrrole immediately via the deprotonation and regenerates Au catalyst (Scheme 1).

To study the reaction mechanism, the deuterium exchange experiments were carried out (Eq. 4). Moderate proportion of deuterated product **2n** observed with deuterium-labelled methanol suggests that the proton sources are supplied not only from methanol but also from the proton released from the intermediate **A** (Scheme 1). A similar deuteration was also observed when the reaction proceeded in either polar or apolar solvent (Eq. 4). The fast intramolecular protonation process is supported further by the fact that we failed to trap the intermediate **B** with different electrophiles. These results suggest that gold(1)-activation of alkyne cycloisomerization might be the rate-determining step.



In summary, we have developed a new and simple pathway to form 2, 5-disubstitute pyrroles in high yields with gold(1)-catalyzed efficient cycloisomerization of acetylenylaziridines. The presence of protic species accelerates the reaction rate and improves the yields of pyrrole products. Further studies on the acetylenylaziridines transformations using Au catalyst and the applications in organic synthesis are underway.

Acknowledgements

Financial support from the National Natural Science Foundation of China (20532050, 20672130, 20821002), the Major Basic Research Development Programme (2006CB806100), Chinese Academy of Sciences and Science and Technology Commission of Shanghai Municipality is acknowledged.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2009.05.091.

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 The reaction of 1j afforded 40% yield of 2j by using 10 mol % of PtCl₂ as catalyst
- at 40 °C. But **1a** failed to react under PtCl₂.
 12. When AgBF₄ was used, the reaction was complicated; **2a** was provided in 55% yield if AgSbF₆ was added.
- The reaction afforded 63% yield of 2a in 29 h with AcOH (0.1 mL) as additive while it gave 26% yield of 2a in 53 h and recover 51% 1a with H₂O (0.1 mL) as additive.